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COVER PICTURE

The cover picture shows the characteristic crystal form of elemental bismuth in the background and a perspective view of the crystal structure $\text{Bi}(\text{Bi}_9)[\text{NbCl}_6]_3$ containing three-capped trigonal prismatic Bi_9^{5+} clusters. The Bi^+ cations in the channels made up of the octahedral halometalate anions are disordered and can be treated in different ways in the crystallographic description. A remarkable feature of the compound is the paramagnetic behaviour, which shows the presence of Nb in the less common oxidation state +IV. Details are discussed in the article by J. Beck and T. Hilbert on p. 2019 ff.



SHORT COMMUNICATIONS

Contents

1985 A. Bacchi,* E. Bosetti, M. Carcelli,
 P. Pelagatti, D. Rogolino

The Structural Role of the Triflate Anion in the Non-Covalent Silver Polymer $[\text{Ag}(\text{LOH})_2-(\text{CF}_3\text{SO}_3)(\text{CH}_3\text{CN})]$ {LOH = α -(4-Pyridyl)-benzhydrol}

Keywords: Crystal engineering / Non-covalent interactions / Hydrogen bonds / Molecular tectonics / Silver compounds

